

## C.6 The nitrogen and phosphorus cycles



Essential idea: Soil cycles are subject to disruption.

# Nitrogen Cycle

## Key Chemical Ingredient: amino acids/proteins

- Earth's atmosphere 80% nitrogen; unavailable to plants; cannot assimilate
- Nitrogen available to plants as
  1. Ammonia ( $\text{NH}_3$ )
  2. Ammonium ( $\text{NH}_4^+$ )
  3. Nitrate ( $\text{NO}_3^-$ )
- Bacteria are essential to the nitrogen cycle
- Nitrogen gas in the atmosphere is very abundant, but is such a stable molecule that bacteria are needed to break it apart and this process consumes much energy
- Nitrogen enters ecosystems by atmospheric deposition (5-10%) or Nitrogen fixation
- $\text{NH}_4^+$  &  $\text{NO}_3^-$  added to soil; dissolved in rain or fine dust (particulates)



The Nitrogen Cycle

# Steps in Nitrogen Cycle

- Five steps are involved in the nitrogen cycle
  1. **Nitrogen fixation** Nitrogen must be fixed in order to be used by plants, its atmospheric form (***Azotobacter***).
  2. **Ammonification** Ammonia ( $\text{NH}_3$ ) is made by decomposing bacteria (***Azotobacter***).
  3. **Nitrification** For those plants who refuse to settle with ammonia, they undergo nitrification. Bacteria (***Nitrobacter***) convert most of the ammonia in soil to nitrite ions ( $\text{NO}_2^-$ )
  4. **Assimilation** This is when plants absorb the substances dropped off by nitrogen fixation and nitrification.
  5. **Denitrification** If the nitrate ions choose not to assimilate they leave the soil and are converted by specialized anaerobic bacteria (***Paracoccus***) in water-logged soil, swamps, lakes.